

David J. Watkin,<sup>a\*</sup> Loren L. Parry,<sup>a</sup> David J. Hotchkiss,<sup>b</sup> Vanessa Eastwick-Field<sup>c</sup> and George W. J. Fleet<sup>b</sup>

<sup>a</sup>Chemical Crystallography, Chemistry Research Laboratory, University of Oxford, Oxford, OX1 3TA, England, <sup>b</sup>Department of Organic Chemistry, Chemical Research Laboratory, Oxford University, Mansfield Road, Oxford OX1 3TA, England, and <sup>c</sup>CMS Chemicals, 9 Milton Park, Abingdon, Oxon, OX14 4RR, England

Correspondence e-mail:  
david.watkin@chem.ox.ac.uk

#### Key indicators

Single-crystal X-ray study  
*T* = 150 K  
Mean  $\sigma(\text{C}—\text{C})$  = 0.002 Å  
*R* factor = 0.044  
*wR* factor = 0.074  
Data-to-parameter ratio = 12.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## 3,5-O-Isopropylidene-2-C-methyl-D-xylonolactone

The ring size of both the lactone and the ketal protecting group in the title compound,  $\text{C}_9\text{H}_{14}\text{O}_5$ , have been established by X-ray crystallographic analysis. The crystal structure consists of hydrogen-bonded spirals parallel to the *b* axis.

Received 6 September 2005  
Accepted 12 September 2005  
Online 17 September 2005

### Comment

Almost all carbohydrate scaffolds contain linear carbon chains (Lichtenthaler & Peters, 2004). The two exceptions that do provide carbohydrates with branched carbon chains are (i) the Kiliani reaction on ketoses which provides efficient access to a set of 2-*C*-hydroxymethylaldonolactones (Hotchkiss *et al.*, 2004; Soengas *et al.*, 2005), and (ii) the treatment of sugars with base to give 2-*C*-methyl aldonic acids, also known as saccharinic acids (Bols, 1996). However, the reaction of base with sugars is complex: glucose gives a mixture of more than 50 compounds on treatment with calcium hydroxide, of which branched sugars comprise a very small percentage (Yang & Montgomery, 1996). Better yields are obtained from ketoses; however, even the optimized conditions (several weeks under careful control in a laborious procedure) for treatment of D-fructose with calcium hydroxide afford 2-*C*-methyl-D-ribonolactone in only 11% yield (Whistler & BeMiller, 1963). Very low yields of branched lactones have been isolated from similar treatment of L-sorbose (Ishizu *et al.*, 1972). A further ketohexose, D-tagatose (1), has recently become available in quantity as a new food additive (Skytte, 2002); (1) has the potential for making 2-*C*-methyl-D-xylonolactone as a branched-sugar building block under green environmentally friendly conditions. Treatment of D-tagatose with aqueous calcium hydroxide produces a very complex mixture of products. In order to identify the branched-chain sugar products, it was necessary to make authentic samples of easily crystallized derivatives.



A crystalline acetonide was obtained from treatment of 2-*C*-methyl-D-xylonolactone with acetone in the presence of acid. The absolute stereochemistry of (2) is determined by using D-tagatose (1) as the starting material; however, there are ambiguities in the synthesis with regard to the relative stereochemistry at C-2 of the lactone, the ring size of the lactone and the ring size of the ketal. X-ray crystallographic analysis removed all the ambiguities and firmly established the structure of the acetonide as (2).

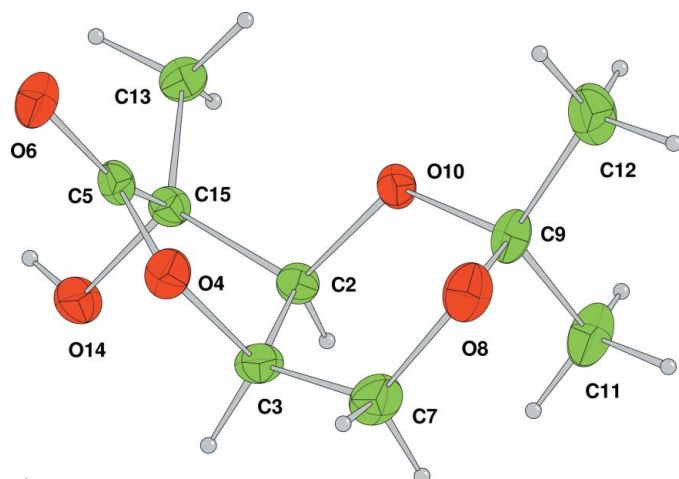


Figure 1

The title compound, with displacement ellipsoids drawn at the 50% probability level. H atoms are shown as spheres of arbitrary radius.

## Experimental

The acetone (2) was prepared as in the *Comment* section and crystallized from ethyl acetate:cyclohexane (m.p. 428–431 K) as long fibrous needles.  $[\alpha]_D^{23} +82.2$  ( $c$  0.67 in  $\text{CHCl}_3$ ).

### Crystal data

$\text{C}_9\text{H}_{14}\text{O}_5$	$D_x = 1.362 \text{ Mg m}^{-3}$
$M_r = 202.21$	Mo $K\alpha$ radiation
Monoclinic, $P2_1$	Cell parameters from 1419 reflections
$a = 8.3764$ (3) Å	$\theta = 5\text{--}30^\circ$
$b = 5.9861$ (2) Å	$\mu = 0.11 \text{ mm}^{-1}$
$c = 10.4690$ (4) Å	$T = 150 \text{ K}$
$\beta = 110.0336$ (12)°	Lath, colourless
$V = 493.17$ (3) Å <sup>3</sup>	$1.00 \times 0.28 \times 0.12 \text{ mm}$
$Z = 2$	

### Data collection

Nonius KappaCCD diffractometer	1549 independent reflections
$\omega$ scans	1549 reflections with $I > -3.0\sigma(I)$
Absorption correction: multi-scan	$R_{\text{int}} = 0.034$
( <i>DENZO/SCALEPACK</i> ;	$\theta_{\text{max}} = 30.0^\circ$
Otwinowski & Minor, 1997)	$h = -11 \rightarrow 11$
$T_{\text{min}} = 0.81$ , $T_{\text{max}} = 0.99$	$k = -8 \rightarrow 8$
7941 measured reflections	$l = -14 \rightarrow 14$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F^2) + (0.04P)^2 + 0.03P]$
$R[F^2 > 2\sigma(F^2)] = 0.044$	where $P = (\max(F_o^2, 0) + 2F_c^2)/3$
$wR(F^2) = 0.074$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 0.94$	$\Delta\rho_{\text{max}} = 0.28 \text{ e Å}^{-3}$
1549 reflections	$\Delta\rho_{\text{min}} = -0.24 \text{ e Å}^{-3}$
127 parameters	
H-atom parameters constrained	

Table 1

Hydrogen-bond geometry (Å, °).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
$\text{O14--H1}\cdots\text{O6}^i$	0.84	2.00	2.837 (2)	176

Symmetry code: (i)  $-x + 1, y - \frac{1}{2}, -z$ .

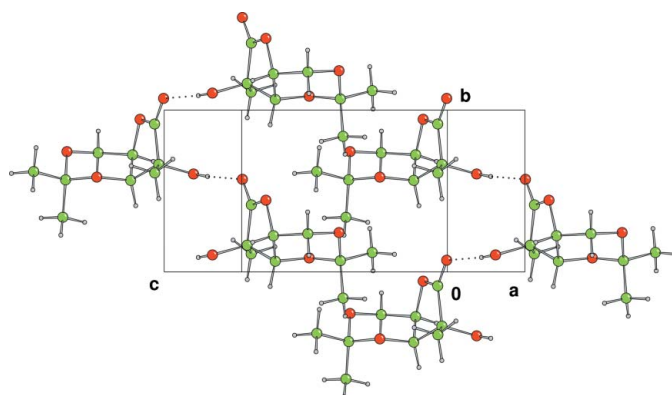


Figure 2

Projection of the structure perpendicular to the  $b$  axis, showing the molecules linked into hydrogen-bonded spirals parallel to  $b$ .

In the absence of significant anomalous scattering, Friedel pairs were merged, and the absolute configuration assigned from the known starting materials.

The H atoms were all located in a difference map, but those attached to C atoms were repositioned geometrically. The H atoms were initially refined with soft restraints on the bond lengths and angles to regularize their geometry (C–H in the range 0.93–0.98 Å and O–H = 0.82 Å) and displacement parameters [ $U_{\text{iso}}(\text{H})$  in the range 1.2–1.5 times  $U_{\text{eq}}$  of the parent atom], after which they were refined with riding constraints.

Data collection: *COLLECT* (Nonius, 1997–2001); cell refinement: *DENZO/SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO/SCALEPACK*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *CRYSTALS*.

Financial support from the EPSRC (to DH) is gratefully acknowledged.

## References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
- Betteridge, P. W., Carruthers, J. R., Cooper, R. I., Prout, K. & Watkin, D. J. (2003). *J. Appl. Cryst.* **36**, 1487.
- Bols, M. (1996). *Carbohydrate Building Blocks*. New York: John Wiley & Sons, Inc.
- Hotchkiss, D., Soengas, R., Simone, M. I., van Ameijde, J., Hunter, S., Cowley, A. R. & Fleet, G. W. J. (2004). *Tetrahedron Lett.* **45**, 9461–9464.
- Ishizu, A., Yoshida, K. & Yamazaki, N. (1972). *Carbohydr. Res.* **23**, 23–29.
- Lichtenthaler, F. W. & Peters, S. (2004). *Compt. Rend. Chim.* **7**, 65–90.
- Nonius (1997–2001). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Skytte, U. P. (2002). *Cereal Foods World*, **47**, 224–227.
- Soengas, R., Izumori, K., Simone, M. I., Watkin, D. J., Skytte, U. P., Soetaert, W. & Fleet, G. W. J. (2005). *Tetrahedron Lett.* **46**, 5755–5759.
- Watkin, D. J., Prout, C. K. & Pearce, L. J. (1996). *CAMERON*. Chemical Crystallography Laboratory, University of Oxford, England.
- Whistler, R. L. & BeMiller, J. N. (1963). *Method Carbohydr. Chem.* **2**, 484–485.
- Yang, B. Y. & Montgomery, R. (1996). *Carbohydr. Res.* **280**, 27–45.